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Accepted in Journal Chromatography Online First 4 May 2012 http://dx.doi.org/10.1016/j.chroma.2012.04.074

## Qualitative analysis of algal secretions with multiple mass spectrometric platforms

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## ABSTRACT

Lipid secretions from algae pose a great opportunity for engineering biofuel feedstocks. The lipid exudates could be interesting from a process engineering perspective because lipids could be collected directly from the medium without harvesting and disrupting cells. We here report on the extracellular secretions of algal metabolites from the strain UTEX 2341 (Chlorella minutissima) into the culture medium. No detailed analysis of these lipid secretions has been performed to date. Using multiple mass spectrometric platforms, we observed around 1000 compounds and were able to annotate 50 lipids by means of liquid chromatography coupled to accurate mass quadrupole time-of-flight mass spectrometry (LC-QTOF), direct infusion with positive and negative electrospray ion trap mass spectrometry and gas chromatography coupled to mass spectrometry (GC-MS). These compounds were annotated by tandem mass spectral (MS/MS) database matching and retention time range filtering. We observed a series of triacylglycerols (TG), sulfoquinovosyldiacylglycerols (SQDG), phosphatidylinositols and phosphatidylglycerols, well betaine lipids diacylglyceryl-N,N,Nas as trimethylhomoserines (DGTS).

## **1. INTRODUCTION**

Algae secrete small molecules such as fatty acids, polysaccharides, small peptides and amino acids as well as exotoxins into their environment [1]. Understanding such

secretions is important for deeper insights into laboratory microecosystems of algae, metabolism studies of algae and algal biofuel research. Pratt (1942) observed such extracellular secretions from Chlorella algae diffuse through cell walls and leak into the culture medium [2]. The algal exudate (Chlorellin) had inhibitory effects on bacteria and algae. The mixture was later analyzed to contain free fatty acids which are cytotoxic to cell membranes [3]. Isotopic tracer experiments were used to postulate that the excretion of a series of amino acids are synchronized to the light cycle of algal photosynthesis [4,5]. Bell and Mitchel (1972) observed that bacteria belonging to the genus Spirillum were attracted within 10-15 minutes to algal exudates from the diatomaceous algae Skeletonema [6]. Bacteria can grow in algal exudates such as amino acids and sugars without additional carbon sources [7]. It also has been discussed that macroalgae release chemical compounds in defense of herbivores that graze on such algae [8] or predators that actively evolved to detect chemical signals released from algae [9]. Toxic algal secretions including domoic acid or brevetoxins occur during harmful algal blooms (HAB) and are threatening to other wildlife species [10]. It has been hypothesized that algae and bacteria release mucoid substances as protection against ultraviolet radiation [11]. On diatom algae (phytoplankton) the extracellular mucilages contain a complex mixture of proteoglycans and sulfated polysaccharides [12].

In this research paper we investigate the extracellular lipid secretions (see **Figure 1**) from the unicellular green algae UTEX 2341 (annotated as *Chlorella minutissima* [13] [14]). Such extracellular secretions were reported previously by Gladu et al. where an scanning electron microscope (SEM) image shows mucoid substances outside the cell walls [14]. UTEX 2341 is a promising candidate for biofuel and biomass production, however the molecular composition of lipid secretions has not been reported in detail. The analysis of such material at the molecular level and the complexity calls for a series of analytical platforms and separation techniques to be used (see **Figure 2**). Fast fingerprint infusion spectra can be obtained with the Direct Analysis in Real Time (DART) ion source coupled to mass spectrometers [15]. Fatty acids analysis can be performed best by converting free fatty acids into their methyl esters (FAME) and analyzing them with GC-MS. Using a preseparation technique such as thin-layer chromatography (TLC) enhances the information content of FAME analysis [16]. Also Matrix-assisted laser desorption/ionization coupled to accurate mass time-of-flight instruments (MALDI-TOF-MS) were used for the analysis of algal lipids in the past [17]. A detailed fragmentation analysis using tandem mass spectrometry analysis software (MS/MS) is possible by nanoelectrospray (nanoESI) chip based direct infusion [18]. Using ultra-high resolving power Fourier Transform Ion Cyclotron Resonance Mass Spectrometry (FT-ICR-MS) it is possible to analyze multiple lipid classes and especially generate unique elemental compositions for lipid species [19]. A detailed separation of multiple lipid species can be obtained with high-performance liquid chromatography coupled to accurate mass quadrupole time-of-flight mass spectrometers (LC-QTOF) [20,21] or LC-Orbitrap mass spectrometry [22]

## **2.** METHODS

## 2.1. Algal growth and separation of algal secretions

The algal strain UTEX 2341 (assigned as *Chlorella minutissima*, also assigned to the *Nannochloropsis* genus [14] and other species [13]) was obtained from the Culture Collection of Algae at the University of Texas at Austin (UTEX). Chlorella UTEX 2341 was maintained in N8Y medium, prepared by adding 0.1% yeast extract to N8 medium at 25°C and irradiated with fluorescent light at ~2000 Lux on a 16:8 h light/dark cycle. For extracellular matter production UTEX 2341 was grown on solid ATCC medium #5 (sporulation agar) in Petri dishes. A control plate (no growth) was included for comparison. After one week a yellow-white secretion from individual UTEX 2341 colonies was observed (see **Figure 1**). The extracellular matter was separated and extracted with a modified Folch method [23]. To do this, the secreted matter was first washed from the algal colony using 2 ml methanol and filtered through a 0.22  $\mu$ m filter into a 50 ml Falcon tube. The total volume was adjusted to 5 ml with methanol and 10 ml chloroform was added and vortexed. Additionally 3 ml water was added and vortexed. The mixture was centrifuged at 1000 g for 2 minutes, and the upper phase, interface and lower phase were collected and lyophilized for 48 hours.

#### 2.2. Direct Analysis in Real Time (DART) analysis

A Direct Analysis in Real Time (DART) ion source (IonSense Inc, Saugus, MA) was coupled to a ThermoFinnigan LTQ linear iontrap (ThermoFinnigan, San Jose CA) operating in full-scan mode from 200-2000 Da. Spectra were recorded with the Xcalibur software. Infusion mass spectra were summed over abundant scans. Solvent blanks were measured to exclude peaks from contaminations. The infusion time was around 5-10 seconds. The DART 100-CE was operated with an effective Helium flow rate of 2.0 l/min. The heater control was set to 250°C, the grid electrode voltage was 250 V, the discharge electrode voltage was 150 V and needle voltage was set to 3000 V.

### 2.3. GC-MS FAME analysis

In order to analyze fatty acids in the GC-MS, their methyl ester analogs (FAME) had to be prepared. The dried sample was dissolved in 200 µl hexane and transferred to another glass vial and mixed with 50 µl toluene, and then 100 µl BCl<sub>3</sub>-methanol was added (Sigma-Aldrich; Supelco 33353; boron trichloride in methanol 12%). The mixture was heated in a water bath at 75°C for 10 min. After cooling, 200 µl distilled water was added. The organic and aqueous phases were then separated and 100 µl of the upper organic phase containing the fatty esters was used for further GC-MS analysis. All solvents and plastic Eppendorf tubes were screened to determine the level of contamination (solvent blanks and method blanks). The injector was an Apex ProSep 800 precolumn separation inlet (Apex Technologies, Inc., NJ). The precolumn mode program was set to splitless mode for 0.17 min, and then changed to GC split for 2.5 min. The precolumn oven of the ProSep was held at 50°C for 0.10 min and then programmed to 250°C at 75°C/min and held for 0.10 min, finally got to 350°C at 50°C/min and held for 32 min. The precolumn method here was for FAME analysis.

The samples were analyzed with an Agilent 6890 gas chromatograph coupled to an Agilent 5973 MSD (Agilent, Santa Clara, CA). 5  $\mu$ l sample was injected at different split ratios of 10:1, 20:1 and 100:1 by an Agilent 7683B series autosampler (Agilent, Santa Clara, CA). The GC column was an Agilent HP5-MS capillary column bonded 5% phenyl, 95% dimethylpolysiloxane (30m×250 $\mu$ m i.d., 0.25 $\mu$ m film thickness). The

temperature program was started at 60°C for 0.5 min. Then the temperature was ramped to 320°C at 10°C/min, and then held at 325°C for 10 min. Helium was used as carrier gas with constant flow rate of 1.0 ml/min. The solvent delay was 6.35 min. The MS filament source temperature was set to 230°C; the quadrupole temperature was set to 150°C and the transfer line temperature was 280°C. Spectra were acquired in positive (70eV) full scan mode from 50-650 *m/z* at 2 spectra/s scan speed.

#### 2.4. Nanoelectrospray chip-based infusion into a linear iontrap mass spectrometer

A chip based nano-electrospray infusion (Advion Nanomate) was coupled to a LTQ linear iontrap mass spectrometer (Thermo Fisher Scientific). 10 µg of the dried Folch extract was resuspended with 100 µl methanol/chloroform (9/1, v/v) (degassed with nitrogen). The sample was vortexed and centrifuged for 2 min at 14,000 rpm. Then 10 µl are diluted with 90 µl methanol/chloroform (9/1, v/v) containing 7.5 mM ammonium acetate and were transferred to the well plates. The plates were covered with aluminum foil to prevent evaporation of the solvent. The infusion of the extracted lipid samples was performed using with an Advion "C" chip, the Nanomate cooling plate was set to 10°C, the Nanomate gas pressure to 0.4 psi and the voltage to 2.0 kV. 10  $\mu$ l were infused with an aspiration delay of 3 seconds. A data-dependent MS/MS method collected tandem mass spectra in positive and negative mode over a range of 20 minutes infusion time. The covered m/z range in the method was split from 350-450 Da, 450-750 Da and 750-1200 Da in order to increase the number of MS/MS spectra for individual components. Each file contained between 100-150 MS/MS precursor ions and the related product ion spectra that averaged between 1-30 scans each. Solvent blanks and method blanks were measured to analyze existing background contaminants from plastic material and extraction solvents.

## 2.5. Chromatographic and mass spectral settings on the Q-TOF instrument

Instrumentation consisted of an Agilent 1200 LC stack (binary pump SL, degasser, autosampler, thermostat, column oven) interfaced to an Agilent 6530 accurate-mass quadrupole time-of-flight (Q-TOF) with a JetStream ESI source. Due to the low solubility

of the excretion material in water/methanol, 5 µl of the Nanomate infusion solvent was used for LC-MS analysis (see method above). The material was injected on an Acquity 1.7 µm BEH HILIC 2.1x150 mm column (Waters Corporation, Milford, MA). Column temperature was kept at 40°C. The mobile phase consisted of water with 5 mM ammonium acetate and 0.2% acetic acid (A) and 9:1 acetonitrile/water with 5 mM ammonium acetate and 0.2% acetic acid (B). The gradient method was: 0-4 min - 100% B, 4-12 min – linear gradient to 45% B, and 12-20 min – 45% B. The column was reequilibrated for 20 min following the separation of each sample, and the flow rate was constant at 0.25 ml/min throughout the gradient method and re-equilibration. MS and MS/MS data were collected with a 0.25 sec scan rate in both profile and centroid modes, and mass calibration was maintained by constant infusion of reference ions at 121.0509 and 922.0098 m/z. MS/MS data was generated utilizing data-dependent MS/MS triggering with dynamic exclusion. Precursor ions, with a minimum 1k signal intensity were isolated with a 4 m/z isolation width (medium setting), and a variable collision energy was applied based on precursor ion m/z (10 eV + 0.03 eV \* ion m/z). Ions were excluded from data dependent MS/MS analysis for 30 sec following acquisition of two spectra. Data were exported into the open exchange format mzXML. Samples were measured in negative and positive mode. Additionally blank runs were acquired to determine the level of background contaminations.

#### 2.6. Data processing and MS/MS database search

The identification of compounds was based on authentic reference MS/MS spectra and an in-house developed MS/MS database (LipidBlast). The QTOF-LC-MS/MS spectra were extracted with Agilent MassHunter software and exported as MGF format (10 counts abundance minimum). For the LTQ instrument the Thermo RAW files were converted using the Thermo ExtractMSn software and the freely available DeconMSn program [24]. The MGF format is a container format that holds multiple MS/MS scans with their referenced precursor ion masses. In order to create consensus MS/MS scans and to reduce the number of spectra all similar MS/MS scans with the same precursor ions were clustered using the MSCluster program [25]

For low-resolution MS/MS search the freely available NIST MS Search GUI program [26] and the NIST MSPepsearch GUI [27] were used. A precursor search accuracy of  $\pm 0.4$  Da and a product ion tolerance to  $\pm 0.8$  Da was set for the low resolution iontrap and  $\pm 0.05$  Da precursor accuracy and  $\pm 0.4$  Da product ion tolerance was applied in case of the high resolution QTOF instrument. A precursor accuracy of 0.008 Da was used for the direct lookup of masses from the QTOF in data-tables containing accurate adduct masses.

All calculations were performed on a Monarch Computer Dual Opteron 254 (2.8 GHz) with an ARECA-1120 Raid-6 array using WD Raptor and Samsung hard disks (max hard disk burst read/write transfer rate 500 MByte/s) equipped with 2.8 GByte RAM running a 32-bit Windows XP. An additional RamDisk (QSoft Ramdisk Enterprise) was used for file based operations allowing burst read-write rates of 1000 MByte/s.

## **3. RESULTS**

#### 3.1. Analysis of algal secretions with DART-linear iontrap mass spectrometry

The DART analysis allows the fast acquisition of mass spectra without any sample preparation or preseparation, because the sample is directly infused by holding a glass capillary into the open DART ion source. Such fingerprinting methods are preferable in order to perform classifications in a high-throughput mode. A complex mixture was detected from the mass spectrum (see **Supplement**) and based on this observation additional analysis steps were performed as discussed below.

## 3.2. Analysis of algal free fatty acid secretions with GC-MS (FAME method)

Two main free fatty acids components were observed in the excretion by FAME analysis. C16:0 (Hexadecanoic acid methyl ester; methyl palmitate) and C18:0 (Octadecanoid acid methyl ester; methyl stearate) were annotated using authentic FAME reference standards with retention time and mass spectral matching. **Figure 3** shows the chromatogram and the mass spectra with the characteristic FAME marker ions m/z 74, 87, 143 and molecular ions m/z 270 (C16:0) and m/z 298 (C18:0). No other free fatty acids as FAMES were observed in the excretion using these characteristic m/z markers.

#### 3.3. Analysis of algal secretions with chip based nanoESI direct infusion

The chip-based nanoelectrospray direct infusion coupled to a low resolution iontrap platform utilized data-dependent MS/MS scans in positive and negative mode (see **Figure 4**). Because of the extended infusion times it is possible to collect MS/MS scans over the range of minutes. Around 2500 MS/MS precursors were fragmented in positive mode and around 1400 in negative mode. Compounds were identified based on precursor search and subsequent MS/MS match. Among the annotated compounds are triacylglycerols, phosphatidylcholines and DGTS betaine lipids in positive mode and sulfoquinovosyldiacylglycerols (SQDG) and phosphatidylinositols in negative mode (see **Table 1**). Due to the complexity of the exudate and same mass isobaric overlaps many of the tandem mass spectra represented mixed compounds.

### 3.4. Analysis of algal secretions with LC-QTOF-MS/MS

The high-resolution accurate mass LC-MS was performed in positive mode and negative mode (see Figure 5). Around 3000 signals (m/z- retention time pairs) were observed in positive mode and 1300 m/z-RT pairs in negative mode. 1400 MS/MS scans in positive and negative mode were acquired. Half of these MS/MS scans were redundant, because they were obtained from the same precursor ion. The positive ionization mode allowed the annotation of triacylglycerols, phosphatidylcholines and diacylglyceryl-N,N,Ntrimethylhomoserines (betaine lipids), whereas in negative mode several phosphatidylinositols and sulfoquinovosyldiacylglycerols were observed (see **Table 1**). All annotations were performed via accurate mass precursor search, retention time matching and additional MS/MS search or confirmation of product ion matches. Because of the high complexity many compounds co-eluted, however MS/MS product ion search and fatty acyl losses allowed for confident annotation of compound classes and fatty acyl carbon chain length and degree of unsaturation. The method did not allow for assignment of stereochemistry and regiospecific or positional analysis of fatty acyl chains. Using negative ion mode a series of fatty acyl chains were observed including m/z 249.153 (C16:3), 255.233 (C16:0), 281.248 (C18:1), 283.264 (C18:0), 295.229 (C18:2(OH)), 297.155 and 297.243 (epoxy/OH/cyclo FA), 311.170 (C20:0), 321.175, 325.185 and 339.200 (epoxy/OH/cyclo FA). The complete elucidation of these fatty acyls requires a more detailed analysis with picolinyl, TMS or DMOX derivates [28]. In addition, an exhaustive query on matching accurate precursor ion masses against lipid databases was performed yielding a high number of tentative lipid annotations (see supplemental data). However, these tentative annotations lack confirmatory MS/MS product spectra and therefore such annotations potentially have a higher false-positive and false negative-rate.

## **4. DISCUSSION**

## 4.1. Taxonomic classification of algae is still a challenging process

Correct taxonomic classifications are important for the compound annotations process. If taxonomy-metabolite relationships are known, such target compound lists can be assembled from databases and a targeted analytical screening process can be performed [29]. The problem is that even in major strain collections, taxonomy annotations may be wrong [13]. The classifications found in these strain collections can be based on microscopic observations, chemotaxonomy data such as the content of alkanes, chlorophyll or carotenoids or they are purely based on genomic information. The algal strain UTEX 2341 (Chlorella minutissima) was assigned to the Nannochloropsis species in one publication [14]. This information is also maintained at the UTEX website (http://www.sbs.utexas.edu/utex/). A phylogenetic comparison based on the 18S ribosomal RNA subunit placed UTEX 2341 (named as Chlorella multissima) into the Trebouxiophyceae class, within close proximity to Botryococcaceae and Chlorellaceae as well as *Nanochlorum eukaryotum* (Nannochloris species) [30]. A very recent sequence comparison based on the 23s rRNA genomic sequence [31], shows a cladogram with UTEX 2341 in near proximity to other Chlorella species [32]. A final correct classification may be obtained after all 4000 Chlorophyta species (green algae) will be fully sequenced.

## 4.2. Algae contain and secrete diverse compound mixtures during their life cycle

Algae contain a series of lipid-like compound classes including free fatty acids, sterols, carotenes, phospholipids, galactolipids, betaine lipids and glycerolipids [33]. The main phospholipid compound classes are phosphatidylcholine (PC), phosphatidylethanolamines (PE), phosphatidic acids (PA), phosphatidylinositols (PI), phosphatidylglycerols (PG). The main glycolipids include monogalactosyldiacylglycerol (MGDG), digalactosyldiacylglycerols (DGDG) and sulfoquinovosyldiacylglycerols (SQDG). Nonpolar lipids include monoacylglycerols (MG), diacylglycerols (DG) and triacylglycerols (TG). Many algae and bacteria also substitute nitrogen for phosphorus and form phosphate-free betaine and ornithine lipids [34]. These betaine lipids include diacylglyceryl-N,N,N-trimethylhomoserine (DGTS). diaclyglycerylhydroxymethyl-N,N,N-trimethyl-beta-alanine (DGTA) and l,2-diacylglyceryl-3-Ocarboxyhydroxymethylcholine (DGCC) [35-38]. In the model algae Chlamydomonas reinhardtii phosphatidylcholine (PC) is completely replaced by the betaine lipid diacylglycerol-O-(N,N,N-trimethyl)-homoserine (DGTS) [39]. Chlorella minutissima contains high contents of DGTS as well as PC as major phospholipids [40]. However the closely related Chlorella vulgaris strain does not contain DGTS but only phosphatidylcholine [36]. DGTS or DGTA were observed in the UTEX 2341 secretions (see Figure 6). Using MS/MS analysis the DGTS and DGTA betaine lipids are easily detected by the dominant product ion m/z 236 [41,42] and DGCC by the ions m/z 104 and 132 [43]. Sulfoquinovosyldiacylglycerols (SQDG) were observed in Chlorella more than 50 years ago [44]. Chlorella algae grown on cysteine as a sole sulfur source instead of sulfate, had a fifteen times lower CO<sub>2</sub> fixation rate, synthesized larger amounts of triglycerides and lost their ability to grow photoautotrophically [45].

With the FAME GC-MS methods we were able to detect the main free fatty acids C16:0 and C18:0 in the algal secretions. Because we observed additional lipid-bound fatty acids with the LC-MS/MS method we have to assume that no transesterification occurred under mild reaction temperatures. A report by Wood et al [46] showed that the fatty acid content of the algae UTEX 2341 heavily fluctuates depending on the carbon and nitrogen source. High C16:0 content was reported under basal ammonia and an increase of 30% for C16:1 was observed under glycerol and ammonia as medium. Another report only

shows minor free fatty acid contents of C16:0 in the algae itself [47]. A detailed fatty acid methyl ester analysis of *Chlorella* species using a transesterification method and GC-MS revealed 160 different fatty acids from C7:0 to C30 chain length, with odd and even carbon numbers, different degrees of unsaturation, cyclopropane- and methyl-substitutions [48].

We observed no bacterial colonies on the plates using microscopic analysis. However it can not be totally excluded that algae were contaminated and bacteria were feeding on exudate and henceforth that a bacterial strain or community was analyzed. Generally bacterial contamination can be analyzed by performing microscopic analysis [14,49], staining methods or better genomic sequencing of these secretions with subsequent matching against genomic databases or phylogenetic analysis. Nevertheless Gladu et al. observed the same mucoid secretions on the outer cell walls of UTEX 2341 [14]. Furthermore the exudate can consist of additional molecular classes (polysaccharides, glycans) [12] that are not covered with our platform. Such unidentified lipophilic compounds can be co-extracted with the Folch method.

#### 4.3. Multiple analytical platforms are needed for analysis of complex matrices

For engineering purposes global sum parameters such as total lipid content [50] are usually sufficient. However, to understand biochemistry, detailed molecular fingerprints and compound identifications are needed. No single platform can cover the complete metabolic signature of a given species. The GC-MS approach only covers volatile compounds or compounds that can be made volatile, but not complex lipids. The LC-QTOF-MS/MS approach uses ultra-high pressure to perform the separation of many compound classes, but with the current setup (20 minute separation time), even UPLC is not able to separate all components within a single lipid compound class. Many components within a single compound class co-elute and overlap. That does not leave enough time for triggering an exhaustive number of MS/MS scans for subsequent compound identifications. Conversely, use of robotic nanoESI infusion allows 20-30 minutes infusion time with only 5-10 microliters of material consumption. This approach allows the collection of hundreds of MS/MS scans. One severe drawback is that isobaric compounds may undergo fragmentation due to the selected 1-2 Da precursor windows. That can lead to mixed product ion spectra and false positive identifications, even on ultra-high resolution instruments.

Our current platform collection used here does not investigate higher molecular species such as carbohydrates, proteoglycans or species with a molecular weight higher than 2000 Da. Such species must be covered by carbohydrate sequencing methods or glycomics approaches. For simple fingerprinting of hydrocarbons and carbohydrates pyrolyzer-gas chromatography can be used [51]. Also compounds that do not ionize easily under electrospray conditions are not covered here. Other detection modes such as ultra violet detection (UV), matrix assisted laser desorption (MALDI), evaporative light scattering (ELSD) or atmospheric pressure ionization (APCI) detectors may be used for such purposes. The correct stereochemical assignments are missing from our compound annotations. A completed structural elucidation of thousands of compounds can be time-consuming and requires additional analytical approaches [52]. For better chromatographic resolution and a more detailed analysis a longer UPLC run time would be needed.

Even ultra-high resolution FT-ICR MS with up to one million resolving power would not be totally sufficient to resolve complex species from multiple adducts. The sodiated species PC 35:0 [M+Na]<sup>+</sup> and PE 38:0 [M+Na]<sup>+</sup> have both the same isotopic mass of 798.598900 Da. A direct infusion method would not distinguish both species and the product ion spectra would contain mixed spectra of both species, however these compounds could be easily separated by chromatography. Another issue for LC-MS and direct infusion methods is the selected precursor isolation width for the MS/MS scans which is usually between 1 and 3 Da. All ions in this mass isolation window are fragmented together, resulting in mixed spectra. Also triple-quadrupole mass spectrometry which is often applied for lipid profiling is prone to false-positive and falsenegative identifications [53] unless an additional production ion scan is triggered for MS/MS confirmation (on hybrid or QTRAP instruments) [52]. For a detailed analysis of complex algae lipids the classical methods of thin-layer chromatography (TLC) with subsequent GC-MS FAME analysis [16,17,38] seems to be favorable for the discovery of new and existing compound classes. Once a separation into different compound classes is established, high-resolution mass spectrometry can be used to annotate compounds without the problem of isobaric overlap [19]. The use of accurate mass during the precursor filtering process in MS/MS search lowers the number of false-positive candidates and increases the confidence level in compound annotations.

# 4.4. Tandem mass spectral libraries (MSMS) are currently limited in size and molecular diversity

The identification of small molecules with tandem mass spectral libraries is currently hindered by the small size of the MS/MS libraries. The catalog of commercially available substances (CHEMCATS) list around 16 million chemicals. PubChem currently covers 30 million small molecules. In comparison the largest MS/MS database NIST11 currently covers 5,843 compounds and their representative 95,000 MS/MS spectra with different adduct types obtained under different fragmentation voltages. The public MassBank Service (<u>www.massbank.jp</u>) covers around 15,000 MS/MS spectra from 4,000 compounds [54]. The public RIKEN MSn Spectral Database for Phytochemicals (RESPECT DB, http://spectra.psc.riken.jp/) [55,56] includes around 9,000 MS/MS spectra. The Agilent METLIN Personal Compound Database and Library contains Q-TOF MS/MS reference spectra for 2,278 compounds and the online METLIN database (http://metlin.scripps.edu/) currently contains 28,329 MS/MS spectra from 5,004 metabolites [57]. One possible approach to increase the coverage is the creation of insilico MS/MS libraries that contain computer generated fragments and abundances (see Figure 6), similar to known approaches for peptide identification [52]. It is also useful to enrich such MS/MS databases with taxonomic information in order to allow a rediscovery of unidentified compound spectra.

# 4.5. Pathway databases and compound databases currently lack detailed molecular lipid species for most algal species

The metabolome coverage for less investigated species such as the strain UTEX 2341 in pathway databases is currently none existent. Even one of the best researched algae

(*Chlamydomonas reinhardtii*) lacks the coverage of molecular lipids species across all pathway or metabolite databases. That includes highly curated databases such as the ChlamyDB (http://www.chlamy.org), ChlamyCyc (http://chlamyto.mpimp-golm.mpg.de) and KEGG DB (<u>http://www.genome.jp/kegg/</u>). The KEGG database has incorporated the LipidMaps categories, but lacks the majority of molecular lipids. For each compound class the combination of different fatty acids would expand to hundreds of compounds with varying fatty acyl chains (n=4 to n=50) and different double bond variations [58]. Many single molecular species such as the betaine lipid DGTS(16:0/18:3) or the sulfolipid SQDG(16:0/16:0) are not covered, even they were reported in the model algae Chlamydomonas already more than 20 years ago [39].

The reason of such low coverage is not the quality of the databases itself, but the intrinsic way such molecules are reported. The most common way was to report molecules in publications; however the loss of data is tremendous because it requires optical chemical structure recognition. We discussed in a recent report that molecules, taxonomy and meta-data must be directly submitted to electronic databases to allow collection and later queries [29]. However no such large public repositories currently exist. The National Science Foundation (US) and the Japan Science and Technology Agency (JST) recently acknowledged the lack of metabolome coverage and announced a joint program in 2011 called "Metabolomics for a Low Carbon Society (METABOLOMICS)" to 1) capture all major metabolites, 2) develop of standards and annotations of unknown metabolites 3) identify specialized metabolites of potential value from plants, algae and bacteria.

## 6. CONCLUSIONS

We identified polar lipid compounds from the outer cell wall secretions from the algal strain UTEX 2341 by employing chromatographic separation techniques coupled to high-resolution and low-resolution tandem mass spectrometry (MS/MS). We annotated around 50 compounds using accurate mass precursor search and subsequent MS/MS mass spectral database search. Some of the annotated compounds were triacylglycerols, phosphatidylcholines, betaine lipids and phosphatidylinositols as well as sulfoquinovosyldiacylglycerols. Many of the estimated 600-1000 compounds remain

unknown and require more detailed pre-fractionation steps. The discovery of these substances is challenging due to missing coverage in compound repositories and their absence in mass spectral MS/MS databases.

## ACKNOWLEDGMENTS

The authors thank Hongyun Gou at the Department of Biological and Agricultural Engineering at UC Davis for preparation of the algae cultures. Funding was provided by NSF MCB 1139644, NIH R01 DK078328, NIH RC2 GM092729, NIH ES013932 and by an Agilent Technologies Foundation grant.

## **AUTHOR CONTRIBUTIONS**

Y.C. and J.S.V. and O.F. designed the experiments of algal culture and extracellular matter harvest. Y.C. grew the algae and extracted the extracellular material. J.K.M, D.Y., F.N., A.V. and T.K. performed mass spectrometric experiments. T.K., J.K.M and D.Y. analyzed the results. T.K. performed the mass spectral interpretation and compound annotations. T.K. and O.F. wrote the manuscript in interaction with all contributing authors.

## **COMPETING INTERESTS STATEMENT**

The authors declare no competing financial interests.

## **SUPPLEMENT**

**Supplement 1**: All LC-MS/MS (MZXML) and infusion-MS/MS files (MGF) as well as GC-MS files (netCDF format) and DART-MS and detailed result files (XLS) are publicly available. See: <u>http://fiehnlab.ucdavis.edu/projects/UTEX-2341/</u>

## **FIGURES**



Figure 1: The algae UTEX 2341 (*Chlorella minutissima*) secretes lipid material into the surrounding media. The photo shows a circular yellow secretion around the green cell body. The exudate was collected, extracted and analyzed with different analytical techniques. Bacterial contaminations were not observed by means of a microscopy method.



**Figure 2: Analytical platform to investigate extracellular lipid secretions from algae.** Four different mass spectrometric platforms were used to analyze a complex mixture of lipids.



**Figure 3: A GC-MS fatty acid methyl ester (FAME) analysis revealed two major peaks in the secretions.** These peaks in the chromatogram were identified as C16:0 (hexadecanoic acid methyl ester) and C18:0 (octadecanoid acid methyl ester) using retention time and mass spectral matching.



Figure 4: Chip based nanoelectrospray infusion allows for long direct infusion and data dependent MS/MS. Complex lipids were observed in the algal secretions. The tandem mass spectra are then searched in MS/MS libraries.



**Figure 5: Hydrophilic interaction chromatography (HILIC-UPLC) coupled to high resolution QTOF-MSMS reveals a complex mixture of compounds.** A series of lipids are annotated using MS/MS scans including triacylglycerols (TG), sulfoquinovosyl-diacylglycerols (SQDG), phosphatidylinositols (PI) and phosphatidylcholines (PC).



**Figure 6**: Left panel: Low resolution iontrap MS/MS spectrum from precursor 736.61 Da was identified in a MS/MS library search as the betaine lipid DGTS 34:2;  $[M+H]^+$ ; DGTS(16:0/18:2). Right panel: High resolution QTOF-MS/MS, the substance with precursor 850.7866821 Da was identified as TG 50:1;  $[M+NH4]^+$ ; TG(16:0/16:0/18:1). Precursor mass accuracy was found as -0.00079 Da (0.93 ppm).

## TABLES

**Table 1:** Identified compounds in UTEX 2341 lipid secretions using LC-QTOF-MS/MS and direct infusion nanoESI-iontrap-MS/MS. All compounds were confirmed with MS/MS library search. The reverse dot product represents the level of confidence. Compound abbreviations see text. Compound annotations without retention time were obtained from the low resolution direct infusion method. Doublet identifications from multiple adducts or platforms were removed.

Num	RT [min]	Prec <i>m/z</i> experiment	Prec <i>m/z</i> library	Delta <i>m/z</i>	Rev-Dot library	Library MS/MS match	Platform
1	1.51	793.51996	793.51358	-0.00638	996	SQDG 32:0; [M-H]-; SQDG(16:0/16:0)	LC-MS, Inf-MS/MS
2	1.54	765.48730	765.48228	-0.00502	821	SQDG 30:0; [M-H]-; SQDG(14:0/16:0)	LC-MS, Inf-MS/MS
3	2.13	900.80153	900.80153	0.00000	806	TG 54:4; [M+NH4]+; TG(18:1/18:1/18:2)	LC-MS/MS
4	2.13	900.80153	900.80153	0.00000	706	TG 54:4; [M+NH4]+; TG(18:0/18:2/18:2)	LC-MS/MS
5	2.13	900.80153	900.80153	0.00000	768	TG 54:4; [M+NH4]+; TG(18:0/18:1/18:3)	LC-MS/MS
6	2.16	874.78714	874.78589	-0.00125	837	TG 52:3; [M+NH4]+; TG(16:0/18:1/18:2)	LC-MS/MS
7	2.16	874.78714	874.78589	-0.00125	823	TG 52:3; [M+NH4]+; TG(16:1/18:1/18:1)	LC-MS/MS
8	2.16	874.78714	874.78589	-0.00125	431	TG 52:3; [M+NH4]+; TG(16:0/18:0/18:3)	LC-MS/MS
9	2.19	902.81833	902.81717	-0.00116	662	TG 54:3; [M+NH4]+; TG(18:0/18:1/18:2)	LC-MS/MS
10	2.38	848.77054	848.77025	-0.00029	812	TG 50:2; [M+NH4]+; TG(16:0/16:1/18:1)	LC-MS/MS
11	2.38	848.77054	848.77025	-0.00029	787	TG 50:2; [M+NH4]+; TG(16:0/16:0/18:2)	LC-MS/MS
12	2.38	848.77054	848.77025	-0.00029	642	TG 50:2; [M+NH4]+; TG(16:1/16:1/18:0)	LC-MS/MS
13	2.55	835.53741	835.53368	-0.00373	380	PI 34:1; [M-H]-; GPIns(16:0/18:1)	LC-MS, Inf-MS/MS
14	2.58	833.52075	833.51800	-0.00275	509	PI 34:2; [M-H]-; GPIns(16:0/18:2)	LC-MS/MS
15	2.62	850.78668	850.78589	-0.00079	687	TG 50:1; [M+NH4]+; TG(16:0/16:0/18:1)	LC-MS/MS
16	2.62	850.78668	850.78589	-0.00079	621	TG 50:1; [M+NH4]+; TG(16:0/16:1/18:0)	LC-MS/MS
17	2.62	850.78668	850.78589	-0.00079	776	TG 50:1; [M+NH4]+; TG(14:0/18:0/18:1)	LC-MS/MS
18	2.98	876.80280	876.80153	-0.00127	667	TG 52:2; [M+NH4]+; TG(16:1/18:0/18:1)	LC-MS/MS
19	2.98	876.80280	876.80153	-0.00127	772	TG 52:2; [M+NH4]+; TG(16:0/18:0/18:2)	LC-MS/MS
20	3.32	904.83202	904.83281	0.00079	602	TG 54:2; [M+NH4]+; TG(16:0/18:1/20:1)	LC-MS/MS
21	3.32	904.83202	904.83281	0.00079	745	TG 54:2; [M+NH4]+; TG(18:0/18:1/18:1)	LC-MS/MS
22	3.32	904.83202	904.83281	0.00079	897	TG 54:2; [M+NH4]+; TG(16:0/18:2/20:0)	LC-MS/MS
23	3.32	904.83202	904.83281	0.00079	641	TG 54:2; [M+NH4]+; TG(18:0/18:0/18:2)	LC-MS/MS
24	3.32	904.83202	904.83281	0.00079	531	TG 54:2; [M+NH4]+; TG(16:1/18:1/20:0)	LC-MS/MS
25	3.32	904.83202	904.83281	0.00079	552	TG 54:2; [M+NH4]+; TG(16:1/18:0/20:1)	LC-MS/MS
26	3.32	904.83202	904.83281	0.00079	339	TG 54:2; [M+NH4]+; TG(16:0/18:0/20:2)	LC-MS/MS
27	3.47	902.81833	902.81717	-0.00116	678	TG 54:3; [M+NH4]+; TG(18:1/18:1/18:1)	LC-MS/MS
28	3.47	902.81833	902.81717	-0.00116	919	TG 54:3; [M+NH4]+; TG(16:1/18:0/20:2)	LC-MS/MS
29	3.47	902.81833	902.81717	-0.00116	329	TG 54:3; [M+NH4]+; TG(16:1/18:1/20:1)	LC-MS/MS
30	3.72	876.80280	876.80153	-0.00127	809	TG 52:2; [M+NH4]+; TG(16:0/18:1/18:1)	LC-MS/MS
31	3.72	876.80280	876.80153	-0.00127	827	TG 52:2; [M+NH4]+; TG(16:1/16:1/20:0)	LC-MS/MS
32	3.11	736.60925	736.60912	-0.00013	983	DGTS 34:2; [M+H]+; DGTS(16:0/18:2)	LC-MS/MS
33	3.16	738.62463	738.62477	0.00014	986	DGTS 34:1; [M+H]+; DGTS(16:0/18:1)	LC-MS/MS
34	3.17	764.63788	764.64042	0.00254	990	DGTS 36:2; [M+H]+; DGTS(18:1/18:1)	LC-MS/MS
35	3.22	762.62387	762.62477	0.00090	987	DGTS 36:3; [M+H]+; DGTS(18:1/18:2)	LC-MS/MS
36	9.31	786.60181	786.60126	-0.00055	614	PC 36:2; [M+H]+; GPCho(18:1/18:1)	LC-MS/MS
37	9.34	784.58606	784.58564	-0.00042	600	PC 36:3; [M+H]+; GPCho(18:1/18:2)	LC-MS/MS
38	9.35	760.58679	760.58564	-0.00115	606	PC 34:1; [M+H]+; GPCho(16:0/18:1)	LC-MS/MS
39	9.42	758.56970	758.56995	0.00025	504	PC 34:2; [M+H]+; GPCho(16:0/18:2)	LC-MS/MS
40	11.14	522.35629	522.35596	-0.00033	318	LPC 18:1; [M+H]+; PC(18:1(11E)/0:0)	LC-MS/MS
41	NA	814.950	814.633	-0.317	995	PC 38:2; [M+H]+; GPCho(18:1/20:1)	Inf-MS/MS
42	NA	837.320	837.549	0.229	761	PI 34:0; [M-H]-; GPIns(16:0/18:0)	Inf-MS/MS
43	NA	851.420	851.565	0.145	488	PI 35:0; [M-H]-; GPIns(16:0/19:0)	Inf-MS/MS
44	NA	865.412	865.581	0.169	161	PI 36:0; [M-H]-; GPIns(16:0/20:0)	Inf-MS/MS
45	NA	881.427	881.518	0.091	246	PI 38:6; [M-H]-; GPIns(16:0/22:6)	Inf-MS/MS
46	NA	737.060	737.451	0.391	853	SQDG 28:0; [M-H]-; SQDG(12:0/16:0)	Inf-MS/MS
47	NA	752.130	752.640	0.510	780	DGTS 35:1; [M+H]+; DGTS(16:0/19:1)	Inf-MS/MS
48	NA	754.260	754.656	0.396	773	DGTS 35:0; [M+H]+; DGTS(16:0/19:0)	Inf-MS/MS
49	NA	896.060	896.770	0.710	913	TG 54:6; [M+NH4]+; TG(18:2/18:2/18:2)	Inf-MS/MS
50	NA	895.320	894.755	-0.565	987	TG 54:7; [M+NH4]+; TG(18:2/18:2/18:3)	Inf-MS/MS

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